Presented By Henderson - Kerr Lake Regional Water System

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Meeting the Challenge

Once again we are proud to present our annual drinking water report, covering all drinking water testing performed between January 1 and December 31, 2015. Over the years, we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best-quality drinking water to your homes and businesses. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all of our water users.

Please remember that we are always available to assist you, should you ever have any questions or concerns about your water.

Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about

drinking water from their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or http:// water.epa.gov/drink/hotline.



Community Participation

You are invited to attend the City of Henderson Council Meetings. They are held on the second Monday of each month, beginning at 6 p.m. at City Hall, 134 Rose Avenue, Henderson, NC.

Substances That Could Be in Water

To ensure that tap water is safe to drink, the U.S. EPA prescribes regulations limiting the amount of certain contaminants in water provided by public water systems. U.S. Food and Drug Administration regulations establish limits for contaminants in bottled water that must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of these contaminants does not necessarily indicate that the water poses a health risk.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals, in some cases, radioactive material, and substances resulting from the presence of animals or from human activity. Substances that may be present in source water include:

Microbial Contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, or wildlife;

Inorganic Contaminants, such as salts and metals, which can be naturally occurring or may result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

Pesticides and Herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

Organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production and may also come from gas stations, urban stormwater runoff, and septic systems;

Radioactive Contaminants, which can be naturally occurring or may be the result of oil and gas production and mining activities.

For more information about contaminants and potential health effects, call the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

Failure in Flint

The national news coverage of water conditions in Flint, Michigan, has created a great deal of confusion and consternation over the past year. The water there has been described as being corrosive; images of corroded batteries and warning labels on bottles of acids come to mind. But is corrosive water necessarily bad?

Corrosive water can be defined as a condition of water quality that will dissolve metals (iron, lead, copper, etc.) from metallic plumbing at an excessive rate. There are a few contributing factors but, generally speaking, corrosive water has a pH of less than 7; the lower the pH, the more acidic, or corrosive, the water becomes. (By this definition, many natural waterways throughout the country can be described as corrosive.) While all plumbing will be somewhat affected over time by the water it carries, corrosive water will damage plumbing much more rapidly than water with low corrosivity.

By itself, corrosive water is not a health concern; your morning glass of orange juice is considerably more corrosive than the typical lake or river. What is of concern is that exposure to elevated levels of the dissolved metals in drinking water increases adverse health risks. And there lies the problem.

Public water systems are required to maintain their water at optimal conditions to prevent it from reaching corrosive levels. Rest assured that we routinely monitor our water to make sure what happened in Flint never happens here. For more information on how corrosivity affects water quality, download this informative pamphlet: www.pubs.ext.vt.edu/442/442-665/442-665_PDF.pdf.



Water Main Flushing

Distribution mains (pipes) convey water to homes, businesses, and hydrants in your neighborhood. The water entering distribution mains is of very high quality; however, water quality can deteriorate in areas of the distribution mains over time. Water main flushing is the process of cleaning the interior of water distribution mains by sending a rapid flow of water through the mains.

Flushing maintains water quality in several ways. For example, flushing removes sediments like iron and manganese. Although iron and manganese do not themselves pose health concerns, they can affect the taste, clarity, and color of the water. Additionally, sediments can shield microorganisms from the disinfecting power of chlorine, contributing to the growth of microorganisms within distribution mains. Flushing helps remove stale water and ensures the presence of fresh water with sufficient dissolved oxygen and disinfectant levels, and an acceptable taste and smell.

During flushing operations in your neighborhood, some short-term deterioration of water quality, though uncommon, is possible. You should avoid tap water for household uses at such times. If you do use the tap, allow your cold water to run for a few minutes at full velocity before use, and avoid using hot water, to prevent sediment accumulation in your hot water tank.

Please contact us if you have any questions or if you would like more information on our water main flushing schedule.

MCL Exceedance for TTHMs

For the first quarter of 2015, Henderson-Kerr Lake Regional Water was in violation of the Total Trihalomathanes (TTHMs) MCL allowed by North Carolina's Rules Governing Public Water Systems.

The highest TTHMs locational running annual average (LRAA) concentration of the water samples collected during the period ending March 31, 2015, was 99 parts per billion (ppb), which exceeds the established MCL of 80 ppb. This happened at location code B01 due to low flow/low turnover in the water, which has been corrected with an automatic flusher established in this area.

The highest TTHMs LRAA concentration of the water samples collected during the period ending June 30, 2015 was 86 ppb at location code B03. This happened for the same reason of low flow/low turnover in the water and has also been corrected with an automatic flusher. We mailed out public notices on June 2, 2015, and included them in our water billing during the month of June for both of these violations.

Some people who drink water containing trihalomethanes in excess of the MCL over many years may experience problems with the liver, kidneys, or central nervous systems, and may have an increased risk of getting cancer.

QUESTIONS?

For more information about this report, or for any questions relating to your drinking water, please call Christy Lipscomb, Kerr Lake Regional Water Plant Director, at (252) 438-2141.

Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at www.epa.gov/lead.



Source Water Assessment

The North Carolina Department of Environment L and Natural Resources (DENR), Public Water Supply (PWS) Section, Source Water Assessment Program (SWAP) conducted assessments for all drinking water sources across North Carolina. The purpose of the assessments was to determine the susceptibility of each drinking water source (well or surface water intake) to Potential Contaminant Sources (PCSs). The results of the assessment are available in SWAP Assessment Reports that include maps, background information, and a relative susceptibility rating of Higher, Moderate, or Lower. The relative susceptibility rating of each source for Henderson-Kerr Lake Regional Water was determined by combining the contaminant rating (number and location of PCSs within the assessment area) and the inherent vulnerability rating (i.e., characteristics or existing conditions of the well or watershed and its delineated assessment area).

A Source Water Assessment Plan (SWAP) is available at our office. This plan is an assessment of the delineated area around our listed sources through which contaminants, if present, could migrate and reach our source water. It also includes an inventory of potential sources of contamination within the delineated area, and a determination of the water supply's susceptibility to contamination by the identified potential sources.

According to the Source Water Assessment Plan dated 6/18/14, our water system had a susceptibility rating of Moderate. It is important to understand that a susceptibility rating of Higher or Moderate does not imply poor water quality, only the system's potential to become contaminated by PCSs in the assessment areas. If you would like to review the Source Water Assessment Plan, please feel free to contact our office during regular office hours at (252) 438-2141.

Where Does My Water Come From?

The City of Henderson - Kerr Lake Regional Water System customers are fortunate because we enjoy an abundant surface water supply from Kerr Lake.

Community Water Fluoridation

The safety and benefits of fluoride are well documented. For over 70 years, U.S. citizens have benefited from drinking water containing fluoride, leading to better dental health. Drinking fluoridated water keeps the teeth strong and has reduced tooth decay by approximately 25 percent in children and adults.

Over the past several decades, there have been major improvements in oral health. Still, tooth decay remains one of the most common chronic diseases of childhood. Community water fluoridation has been identified as the most cost-effective method of delivering fluoride to all members of the community, regardless of age, educational attainment, or income level.

Nearly all water contains some fluoride, but usually not enough to help prevent tooth decay or cavities. Public water systems can add the right amount of fluoride to the local drinking water to prevent tooth decay.

Community water fluoridation is recommended by nearly all public health, medical, and dental organizations in the U.S. Because of its contribution to the dramatic decline in tooth decay, the Centers for Disease Control and Prevention (CDC) named community water fluoridation one of the greatest public health achievements of the 20th century. (Courtesy of www.cdc.gov/fluoridation)

Sampling Results

During the past year, we have taken hundreds of water samples in order to determine the presence of any radioactive, biological, inorganic, volatile organic, or synthetic organic organic or synthetic organic organic. The tables below show only those contaminants that were detected in the water. The state requires us to monitor for certain substances less often than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

We participated in the 3rd stage of the EPA's Unregulated Contaminant Monitoring Rule (UCMR3) program by performing additional tests on our drinking water. UCMR3 benefits the environment and public health by providing the EPA with data on the occurrence of contaminants suspected to be in drinking water, in order to determine if the EPA needs to introduce new regulatory standards to improve drinking water quality. Contact us for more information on this program.

SUBSTANCE (UNIT OF MEASURE)			YI SAN	EAR MPLED	MCL [MRDL]	MCLG [MRDLG]	AMOUNT DETECTED	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Barium (ppm)		2	.015	2	2	0.0200	0.0200-0.0200	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits	
Chlorine (ppm)		2	.015	[4]	[4]	0.92	0.08-1.85	No	Water additive used to control microbes	
Fluoride (ppm)			2	.015	4	4	0.78	0.78–0.78	No	Erosion of natural deposits; Water additive that promotes strong teeth; Discharge from fertilizer and aluminum factories
Haloacetic Acids [HAAs]– Stage 2 (ppb)			2	.015	60	NA	281	13–38	No	By-product of drinking water disinfection
TTHMs [Total Trihalomethanes]– Location: B01 (ppb)			2	.015	80	NA	99 ¹	41-88	Yes	By-product of drinking water disinfection
TTHMs [Total Trihalomethanes]– Location: B03 (ppb)			2	.015	80	NA	86 ¹	56–88	Yes	By-product of drinking water disinfection
Total Organic Carbon [TOC] ² (removal ratio)			2	.015	ΤT	NA	1.33	1.21–1.48	No	Naturally present in the environment
Turbidity ³ (NTU)			2	.015	TT = 1 NTU	NA	0.09	0.02–0.09	No	Soil runoff
Turbidity (Lowest monthly percent of samples meeting limit)		2	.015	95% of samples < 0.3 NTU	NA	100	NA	No	Soil runoff	
Tap water samples were collected for lead and copper analyses from sample sites throughout the community.										
SUBSTANCE YEAR (UNIT OF MEASURE) SAMPLED		AL	MCLG	AMOUNT DETECTED MCLG (90TH%TILE)		SITES ABOVE AL/TOTAL SITES	5 VIOLATION	VIOLATION TYPICAL SOURCE		
Copper (ppm)	2014		1.3	1.3	0.1	10	0/31	No Corrosion of household plumbing systems; Erosion of natural deposits		
SECONDARY SUBSTANCES										
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED SMC		ICL	MCLG	AMOUNT RANGI DETECTED LOW-HIG		H VIOLATION	TYPICAL SOUR	TYPICAL SOURCE	
Fluoride (ppm)	2015	2	2.0	NA	0.76	0.52-0.9	98 No	Erosion of na	tural deposits;	Water additive that promotes strong teeth; Discharge from fertilizer and aluminum factories
pH (Units)	2015	6.5	-8.5	NA	7.1	7.1–7.	1 No	Naturally occ	urring	
Sulfate (ppm)	lfate (ppm) 2015		50 NA		19.2 19.2–19.2		0.2 No	Runoff/leachi	Runoff/leaching from natural deposits; Industrial wastes	
UNREGULATED SUBSTANCES ⁴										
SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED		AM DET	AMOUNT RANGE DETECTED LOW-HIGH		: н түр	ICAL SOURCE			
Sodium (ppm)	2015		1	15.4 15.4–15.4		5.4 Na	Naturally present in the environment			

¹Locational running annual average.

²Depending on the TOC in our source water, we MUST have a certain percent removal of TOC or must achieve alternative compliance criteria. If we do not achieve that percent removal, there is an alternative percent removal. If we fail to meet the alternative percent removal, we are in violation of a Treatment Technique.

³Turbidity is a measure of the cloudiness of the water. We monitor it because it is a good indicator of the effectiveness of our filtration system. The turbidity rule requires that 95% or more of the monthly samples must be less than or equal to 0.3 NTU.

⁴Unregulated contaminants are those for which the U.S. EPA has not established drinking water standards. The purpose of monitoring unregulated contaminants is to assist the EPA in determining the occurrence of unregulated contaminants in drinking water and whether future regulations are warranted.

Definitions

AL (Action Level): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

LRAA (Locational Running Annual Average): The average of sample analytical results for samples taken at a particular monitoring location during the previous four calendar quarters under the Stage 2 Disinfectants and Disinfection By-Products Rule.

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

MRDL (Maximum Residual Disinfectant Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NA: Not applicable

NTU (Nephelometric Turbidity Units): Measurement of the clarity, or turbidity, of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

ppb (parts per billion): One part substance per billion parts water (or micrograms per liter).

ppm (parts per million): One part substance per million parts water (or milligrams per liter).

removal ratio: A ratio between the percentage of a substance actually removed to the percentage of the substance required to be removed.

SMCL (Secondary Maximum Contaminant Level): SMCLs are established to regulate the aesthetics of drinking water like taste and odor.

TT (Treatment Technique): A required process intended to reduce the level of a contaminant in drinking water.